Linear Irreversible Thermodynamics for Periodically Modulated Systems

Karel.Proesmans@Uhasselt.be

Introduction

- Linear irreversible thermodynamics
- Time-dependent systems
- Work to work conversion





$$F_{\text{Tot}}(t) = F_{\text{Fric}}(t) + F_{\text{Spring}}(t) + F_{\text{Ext}}(t)$$

$$F_{\text{Fric}}(t) = -\gamma \dot{x}(t)$$

$$F_{\text{Spring}}(t) = -kx(t)$$

$$m\ddot{x}(t) = -\gamma \dot{x}(t) - kx(t) + F_{\text{Ext}}(t)$$

$$= 0$$

(Overdamped Limit)

Work and Heat

Overdamped E.O.M.: $\gamma \dot{x}(t) = -kx(t) + F_{Ext}(t)$

Solution:
$$x(t) = \frac{1}{\gamma} \int_0^\infty d\tau \, e^{-\frac{k\tau}{\gamma}} F_{\text{Ext}}(t-\tau)$$

Thermodynamics:

$$\dot{E}(t) = -F_{\text{Spring}}(t)\dot{x}(t)$$
$$\dot{W}(t) = -F_{\text{Ext}}(t)\dot{x}(t)$$
$$\dot{Q}(t) = \dot{E} - \dot{W} = \gamma \dot{x}(t)^2 \ge 0$$

Cyclic process

Periodic driving: $F_{\text{Ext}}(t + \mathcal{T}) = F_{\text{Ext}}(t)$ Steady state: $x(t + \mathcal{T}) = x(t)$ $\Rightarrow \quad \dot{E} = \frac{1}{\mathcal{T}} \int_0^{\mathcal{T}} dt \, \dot{E}(t) = 0$ $\Rightarrow \quad \dot{W} = -\dot{Q} \le 0$

Kelvin statement of the second law:

,There is no process whose only effect is to accept heat from a single heat reservoir and transform it entirely into work.'

Entropy

Definition:
$$\dot{\bar{S}} = \frac{\dot{\bar{Q}}}{T} \ge 0$$

Quasi-static limit: $\dot{\bar{S}} = 0 \Leftrightarrow \dot{x}(t) = 0$, $\forall t$

Forces and fluxes:
$$\dot{\bar{S}} = \sum_i \mathcal{F}_i J_i$$

Linear regime:
$$J_i = \sum_j L_{ij} \mathcal{F}_j \Rightarrow \dot{\bar{S}} = \sum_{i,j} \mathcal{F}_i L_{ij} \mathcal{F}_j$$

Thermodynamic forces and fluxes

Thermodynamic Force: $F_{\text{Ext}}(t) = \mathcal{F}g(t)$

Thermodynamic Flux:

$$\begin{split} \dot{\bar{S}} &= \mathcal{F}J = \frac{\mathcal{F}}{\mathcal{T}} \int_0^{\mathcal{T}} dt \, g(t) \dot{x}(t) \\ \Rightarrow J &= \frac{1}{\mathcal{T}} \int_0^{\mathcal{T}} dt \, g(t) \dot{x}(t) \\ &= \frac{1}{\mathcal{T}\gamma} \int_0^{\mathcal{T}} dt \int_0^{\infty} d\tau \, g(t) \dot{g}(t-\tau) e^{-\frac{k\tau}{\gamma}} \mathcal{F} \end{split}$$

Work to work conversion

Two work-sources: $F_{\text{Ext}}(t) = F_{\text{Ext},1}(t) + F_{\text{Ext},2}(t)$

 $W_1 = \int dt F_{\text{Ext},1}(t) \dot{x}(t) = T\mathcal{F}_1 J_1 \qquad W_2 = \int dt F_{\text{Ext},2}(t) \dot{x}(t) = T\mathcal{F}_2 J_2$

Working regime: $W_1 > 0$, $W_2 < 0$

Efficiency:

$$\dot{\bar{S}} = \mathcal{F}_1 J_1 + \mathcal{F}_2 J_2 \quad \longrightarrow \bar{\eta} = -\frac{W_2}{W_1} = -\frac{\mathcal{F}_2 J_2}{\mathcal{F}_1 J_1} \le 1$$

The Onsager Matrix

Thermodynamic fluxes:

$$J_1 = \frac{1}{\mathcal{T}} \int_0^{\mathcal{T}} dt \, g_1(t) \dot{x}(t)$$

= $\frac{1}{\mathcal{T}\gamma} \int_0^{\mathcal{T}} dt \int_0^{\infty} d\tau \, g_1(t) \left(\mathcal{F}_1 \dot{g}_1(t-\tau) + \mathcal{F}_2 \dot{g}_2(t-\tau)\right) e^{-\frac{k\tau}{\gamma}}$

Onsager matrix:

$$L = \begin{bmatrix} \frac{1}{T\mathcal{T}\gamma} \int_0^{\mathcal{T}} dt \int_0^{\infty} d\tau g_1(t) \dot{g}_1(t-\tau) e^{-\frac{k\tau}{\gamma}} & \frac{1}{T\mathcal{T}\gamma} \int_0^{\mathcal{T}} dt \int_0^{\infty} d\tau g_1(t) \dot{g}_2(t-\tau) e^{-\frac{k\tau}{\gamma}} \\ \frac{1}{T\mathcal{T}\gamma} \int_0^{\mathcal{T}} dt \int_0^{\infty} d\tau g_2(t) \dot{g}_1(t-\tau) e^{-\frac{k\tau}{\gamma}} & \frac{1}{T\mathcal{T}\gamma} \int_0^{\mathcal{T}} dt \int_0^{\infty} d\tau g_2(t) \dot{g}_2(t-\tau) e^{-\frac{k\tau}{\gamma}} \end{bmatrix}$$

Onsager-Casimir symmetry

Time-reversal:

$$F_{\mathrm{Ext},i}(t), g_i(t) \to \tilde{F}_{\mathrm{Ext},i}(t) = F_{\mathrm{Ext},i}(-t), \tilde{g}_i(t) = g_i(-t)$$

Onsager coefficients:

$$\begin{split} \tilde{L}_{ij} &= \frac{1}{\mathcal{T}\gamma} \int_0^{\mathcal{T}} dt \int_{-\infty}^0 d\tau \, g_i(-t) \dot{g}_j(-t+\tau) e^{\frac{k\tau}{\gamma}} \\ &= \frac{1}{\mathcal{T}\gamma} \int_0^{\mathcal{T}} dt \int_{-\infty}^0 d\tau \, g_j(t) \dot{g}_i(t-\tau) e^{\frac{k\tau}{\gamma}} \\ &= L_{ji} \end{split}$$

Time-symmetric driving: $L_{ij} = L_{ji}$

Conclusions

- Linear irreversible thermodynamics can describe time-dependent systems.
- A generalized Onsager-Casimir relation is found.

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